

## **SINTESIS KOMPOSIT $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$ DAN AKTIVITAS FOTOKATALITIKNYA TERHADAP DEGRADASI METILEN BIRU**

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### **INTISARI**

Sintesis  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  untuk degradasi fotokatalitik metilen biru telah dilakukan. Penelitian ini bertujuan untuk mempelajari pengaruh dopan perak dan sulfur dalam modifikasi fotokatalis  $\text{TiO}_2$  sehingga lebih responsif pada sinar tampak serta aktivitas fotokatalitik terhadap degradasi metilen biru. Sintesis  $\text{Fe}_3\text{O}_4$  dilakukan dengan metode kopresipitasi dari prekursor  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  dan  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ . Material  $\text{Fe}_3\text{O}_4/\text{TiO}_2$  dan  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  disintesis dengan metode sol-gel. Titanium tetraisopropoksida (TTIP) digunakan sebagai prekursor dari  $\text{TiO}_2$ . Perak nitrat ( $\text{AgNO}_3$ ) digunakan sebagai sumber dopan perak dan penggunaan tiourea ( $\text{CH}_4\text{N}_2\text{S}$ ) sebagai sumber dopan sulfur. Material hasil sintesis dikarakterisasi dengan spektrofotometer FT-IR, difraktometer sinar-X (XRD), spektrofotometer SR UV-Visibel, *Scanning Electron Microscope* dengan *Energy Dispersive X-Ray* (SEM-EDX), *Transmission Electron Microscope* (TEM) dan *Vibrating Sample Magnetometer* (VSM). Aktivitas fotokatalis dilakukan dengan cara degradasi fotokatalitik terhadap metilen biru dalam reaktor tertutup disertai variasi perlakuan yaitu pH, waktu penyinaran, massa fotokatalis, volum metilen biru, penambahan *enhancing agent* ( $\text{H}_2\text{O}_2$ ), jenis sinar, dan pengujian penggunaan kembali. Penentuan hasil degradasi fotokatalitik dilakukan dengan menggunakan spektrofotometri UV-Visibel.

Hasil penelitian menunjukkan bahwa sintesis  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  berhasil dilakukan. Material  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  memiliki sifat kemagnetan dengan nilai saturasi kemagnetan sebesar 5,33 emu/gram. Keberadaan dopan perak dan sulfur menurunkan energi celah pita sehingga responsif terhadap sinar tampak. Aktivitas fotokatalitik optimum terjadi pada  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  (2,5%|2,5%) dengan nilai energi celah pita sebesar 2,85 eV. Hasil degradasi fotokatalitik 10 mL metilen biru 5 ppm dengan  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  (2,5%|2,5%) mencapai 88,82% di bawah paparan sinar UV dan 99,18% di bawah paparan sinar tampak pada pH 10, waktu paparan selama 120 menit, massa  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  sebesar 10 mg, dan penambahan 0,1 mL  $\text{H}_2\text{O}_2$ . Material  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  bersifat stabil dalam empat kali penggunaan.

Kata kunci: degradasi,  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$ , dopan, metilen biru

## **SYNTHESIS OF $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$ COMPOSITE AND ITS PHOTOCATALYTIC ACTIVITY TOWARDS DEGRADATION OF METHYLENE BLUE**

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### **ABSTRACT**

Synthesis of  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  for degradation of methylene blue has been carried out. The purpose of this research was to study the effect of silver and sulphur dopants into  $\text{TiO}_2$  modification as photocatalyst to be more responsive on visible light and also its photocatalytic activity towards degradation of methylene blue. The synthesis of  $\text{Fe}_3\text{O}_4$  was initiated through coprecipitation method from  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  and  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$  as the precursors. Both  $\text{Fe}_3\text{O}_4/\text{TiO}_2$  dan  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  were synthesized by using sol-gel method. Titanium tetraisopropoxide (TTIP) was used as the precursor of  $\text{TiO}_2$ . Silver nitrate ( $\text{AgNO}_3$ ) was used as silver dopant source and the usage of thiourea ( $\text{CH}_4\text{N}_2\text{S}$ ) as sulphur dopant source. The synthesized materials were characterized by using FT-IR spectrophotometer, X-Ray Diffractometer (XRD), SR UV-Visible spectrophotometer, Scanning Electron Microscope with Energy Dispersive X-Ray (SEM-EDX), Transmission Electron Microscope (TEM) and Vibrating Sample Magnetometer (VSM). The activity of photocatalysts were performed by photocatalytic degradation towards methylene blue inside a closed photoreactor with several variations which are pH, irradiation time, photocatalyst mass, methylene blue volume, addition of enhancing agent ( $\text{H}_2\text{O}_2$ ), types of light, and reusability test.

The result showed that  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  was synthesized successfully. The  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  material showed magnetic properties as the magnetic saturation value was 5,33 emu/gram. The presence of silver and sulphur dopants lower the band gap and tended to be responsive towards visible light exposure. The optimum photocatalytic activity was obtained on  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  (2,5%|2,5%) with the band gap value was 2,85 eV. The photocatalytic degradation result of 10 mL methylene blue 5 ppm reached 88,82% under UV light exposure and 99,18% under visible light exposure at pH 7, 120 minutes irradiation time, 10 mg of  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$ , and addition of 0,1 mL  $\text{H}_2\text{O}_2$ . The  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$  was stable under four times usage.

**Keywords:** degradation,  $\text{Fe}_3\text{O}_4/\text{TiO}_2\text{-Ag/S}$ , dopant, methylene blue