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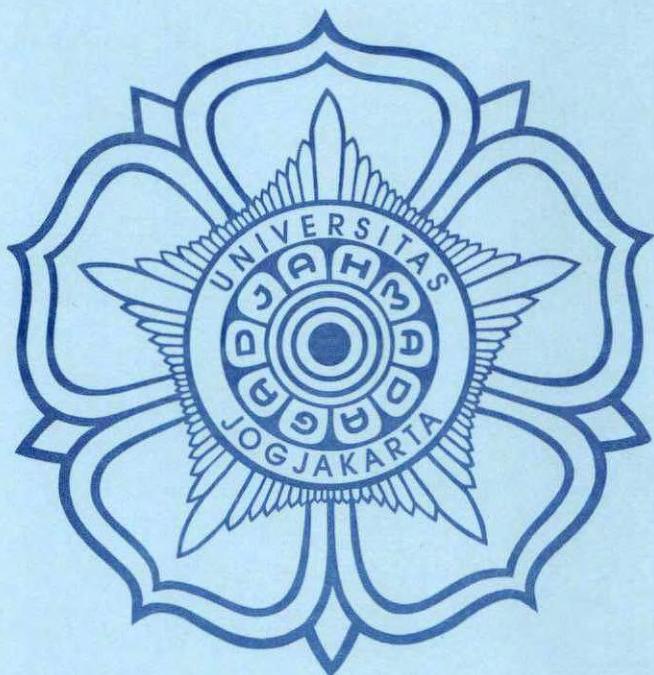
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SINTESIS AMIDA, ESTER DAN ASAM TURUNAN POLLEUGENOL DAN KAJIAN APLIKASINYA UNTUK
PENGOMPLEKS LOGAM

PADA METODE PEMISAHAN EKSTRAKSI CAIR-CAIR DAN TRANSPOR MEMBRAN CAIR RUAH

La Harimu, Prof. Dr. Sabirin Matsjeh

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Lampiran 1. Persentase ion logam yang terekstraksi terhadap variasi pH, rasio mol ligan/logam, waktu ekstraksi dan konsentrasi ion logam menggunakan engeman ion PEAS

a. Variasi pH

Variasi pH	Persentase ion logam yang terekstraksi (%)					
	Fe(III)	Cr(III)	Ni(II)	Co(II)	Cu(II)	Pb(II)
2	85,81					
3	100	63,8	24,4	21,2	65,8	39,1
4	100	77,1	28,4	23,8	78,4	86,7
5	100	85,7	32,3	29,5	85,3	94,7
6	94,3	78,1	29,6	28,2	87,0	100

b. Rasio mol ligan/logam

Rasio mol ligan/logam	Persentase ion logam yang terekstraksi (%)					
	Fe(III)	Cr(III)	Ni(II)	Co(II)	Cu(II)	Pb(II)
2,5	64,1					
4	96,9					
5	100	81,1	28,6	20,9	48,9	81,9
10	100	100	41,3	38,0	88,5	85,6
15	100	96,3	40,5	36,3	92,3	100
20	100	93,2	38,1	34,5	91,0	98,1

c. Rasio mol ligan/logam (awal) terhadap rasio mol ligan/logam (fasa organik)

Rasio mol ligan/logam (awal)	Rasio mol ligan/logam (fasa organik)					
	Fe(III)	Cr(III)	Ni(II)	Co(II)	Cu(II)	Pb(II)
5	4,90	6,71	17,46	70,86	7,34	5,64
10	9,81	9,54	24,17	48,72	10,89	10,80
15	14,71	14,31	36,97	66,81	16,59	13,86
20	19,61	20,47	52,38	115,48	22,52	18,49