

PREPARASI NANOPARTIKEL Fe₃O₄-Cu DAN APLIKASINYA SEBAGAI KATALIS DALAM SINTESIS TURUNAN PROPARGILAMIN MELALUI REAKSI MULTIKOMPONEN

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INTISARI

Penelitian ini bertujuan untuk melakukan preparasi dan mengkarakterisasi nanopartikel Fe₃O₄-Cu serta mengevaluasi aktivitas katalitiknya sebagai katalis asam pada sintesis turunan propargilamin. Penelitian diawali dengan sintesis Fe₃O₄ dengan metode sono-kopresipitasi, dilanjutkan sintesis Fe₃O₄-Cu dengan penambahan logam Cu menggunakan CuSO₄·5H₂O. Hasil sintesis material Fe₃O₄ dan Fe₃O₄-Cu dikarakterisasi menggunakan XRD, FTIR, SEM-EDX, TEM, VSM, SAA, dan uji keasaman secara gravimetri. Uji aktivitas katalitik Fe₃O₄-Cu sebagai katalis asam dalam sintesis propargilamin dilakukan melalui reaksi multikomponen berbahan dasar benzaldehida, fenilasetilena, dan morfolin. Produk reaksi dielusidasi dengan spektrometer inframerah, ¹HNMR, ¹³CNMR, serta *elemental analyzer*.

Hasil analisis FTIR, XRD, dan SEM-EDX menunjukkan bahwa sintesis nanopartikel Fe₃O₄-Cu telah berhasil dilakukan. Katalis Fe₃O₄-Cu yang memiliki nilai keasaman terbesar yaitu Fe₃O₄-Cu 10% mol dengan nilai 6,63 mmol g⁻¹. Analisis TEM menunjukkan bahwa partikel Fe₃O₄-Cu 10% berbentuk bola bulat dan sebagian teraglomerasi. Hasil uji kemagnetan dengan VSM menunjukkan bahwa Fe₃O₄-Cu 10% bersifat superparamagnetit dan dapat dipisahkan dalam medium cair. Hasil analisis SAA mengindikasikan Fe₃O₄-Cu merupakan material mesopori dengan luas permukaan, volume pori, dan diameter pori masing-masing sebesar 50,58 m² g⁻¹, 0,20 cm³ g⁻¹, dan 6,919 nm. Uji aktivitas katalitik menunjukkan bahwa katalis optimum dalam reaksi multikomponen adalah nanopartikel Fe₃O₄-Cu 10%, dimana turunan propargilamin dapat disintesis dengan persen hasil 99%.

Kata kunci: nanopartikel Fe₃O₄-Cu, reaksi multikomponen, sono-kopresipitasi, turunan propargilamin

PREPARATION OF Fe₃O₄-Cu NANOPARTICLE AND AND ITS APPLICATION AS A CATALYST IN MULTICOMPONENT REACTION TOWARDS THE SYNTHESIS OF PROPARGYLAMINE DERIVATIVE

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ABSTRACT

This research was aimed to prepare and to characterize of Fe₃O₄-Cu nanoparticle and to evaluate their catalytic activity as an acid catalyst in the synthesis of propargylamine derivatives. The research was started by synthesizing Fe₃O₄ through a sono-coprecipitation method, followed by the synthesizing Fe₃O₄-Cu through doping of Cu metal using CuSO₄·5H₂O. The materials were characterized by XRD, FTIR, SEM-EDX, TEM, VSM, SAA, and gravimetrically acidity test. The catalytic activity test of Fe₃O₄-Cu as an acid catalyst in synthesis of propargylamine through a multicomponent reaction was carried out from benzaldehyde, phenylacetylene, and morpholine. The products were elucidated with infrared spectrometer, ¹H-NMR, ¹³C-NMR, and elemental analyzer.

The results of FTIR, XRD, and SEM-EDX analysis showed that the synthesis of Fe₃O₄-Cu had been successfully carried out. Fe₃O₄-Cu catalyst has the greatest acidity value at concentration of 10% mol with a value of 6,63mmol g⁻¹. TEM analysis showed that Fe₃O₄-Cu 10% particles were spherical and partially agglomerated. The results of magnetic test with VSM showed that Fe₃O₄-Cu 10% was superparamagnetic and could be separated in liquid medium. The results of SAA analysis indicate that Fe₃O₄-Cu 10% is mesoporous material with surface area, pore volume, and pore diameter of 50,58 m² g⁻¹, 0,20 cm³ g⁻¹, and 6,919 nm, respectively. The catalytic activity test showed that the optimum catalyst in the multicomponent reaction was Fe₃O₄-Cu 10% nanoparticles, where propargylamine derivatives could be synthesized in 99% yields.

Keyword: Fe₃O₄-Cu nanoparticle, multicomponen reaction, propargylamine derivative, sono-coprecipitation