



**SINTESIS Co,N *BIDOPED* ZrTiO₄ SEBAGAI FOTOKATALIS
RESPONSI SINAR TAMPAK POTENSIAL DENGAN VARIASI
KONSENTRASI Co DAN SUHU KAL SINASI**

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INTISARI

Sintesis dan karakterisasi Co,N *Bidoped* ZrTiO₄ sebagai model fotokatalis responsif terhadap sinar tampak telah dilakukan. Tujuan penelitian ini adalah mendapatkan material Co,N *Bidoped* ZrTiO₄ melalui metode sol-gel, mempelajari pengaruh variasi presentase CoSO₄.7H₂O, dan suhu kalsinasi pada material. Sintesis Co,N *Bidoped* ZrTiO₄ dengan metode sol-gel diawali dengan melarutkan titanium isopropoksida (TTIP) dalam etanol absolut yang kemudian direaksikan dengan ZrO₂, nitrogen dari urea dengan konsentrasi 10% (b/b)(N/Ti), dan CoSO₄.7H₂O dengan konsentrasi 0, 1, 3, 5, 7, dan 9% (b/b)(Co/Ti). Variasi konsentrasi CoSO₄.7H₂O di kalsinasi pada suhu 500 °C. 5% Co,N-*Bidoped* ZrTiO₄ di kalsinasi pada variasi suhu 500, 600, 700, 800, dan 900 °C. Material Co,N *Bidoped* ZrO₂ disintesis menggunakan metode refluks sebagai proses kontrol. Semua material tersintesis dikarakterisasi dengan *X-ray diffraction* (XRD), *scanning electron microscopy with energy dispersive X-ray* (SEM-EDX), *fourier transform infrared spectrophotometer* (FTIR), dan *specular reflectance UV-Visible spectrophotometer* (SRUV).

Hasil yang diperoleh menunjukkan bahwa sintesis Co,N *Bidoped* ZrTiO₄ melalui metode sol-gel berhasil dilakukan. Karakterisasi XRD menunjukkan bahwa fase kristal fotokatalis Co,N-*Bidoped* ZrTiO₄ didominasi oleh fase tetragonal dan anatas pada suhu 500 °C. ZrO₂ sebagai *supporting material* mampu meningkatkan kestabilan termal fotokatalis Co,N-*Bidoped* ZrTiO₄. Analisa SEM-EDX menunjukkan bahwa dopan Co dan N berhasil terdoping ke dalam kerangka ZrTiO₄. Spektra IR menunjukkan perubahan intensitas Ti-O. Analisa SRUV menunjukkan bahwa dopan Co dan N mampu menurunkan *band gap* TiO₂. Material dengan penambahan konsentrasi Co 3% dan suhu kalsinasi 500 °C memiliki *band gap* terendah (2.54 eV) dan panjang gelombang tepi absorpsi terpanjang (487 nm).

Kata kunci: Co,N *Bidoped* ZrTiO₄, fotokatalis, sol-gel, suhu kalsinasi, variasi konsentrasi



**SYNTHESIS OF Co,N BIDOPED ZrTiO₄ AS POTENTIAL
VISIBLE-LIGHT RESPONSIVE PHOTOCATALYST WITH VARIOUS Co
CONCENTRATIONS AND CALCINATION TEMPERATURES**

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ABSTRACT

Synthesis and characterization of Co,N Bidoped ZrTiO₄ as a model visible-light responsive photocatalyst has been done. The purpose of this research were to synthesis Co,N bidoped ZrTiO₄ material through sol-gel method, to study the effects of percentage variations of CoSO₄.7H₂O, and calcination temperature on material. Synthesis of Co,N bidoped ZrTiO₄ by sol gel method started with dissolving titanium isopropoxide (TTIP) in absolute ethanol, then reacted with ZrO₂, nitrogen from urea 10% (w/w)(N/Ti) and CoSO₄.7H₂O with various concentration of 0, 1, 3, 5, 7, and 9% (w/w)(Co/Ti) . The variation concentration of CoSO₄.7H₂O was calcined at 500 °C. 5% Co,N Bidoped ZrTiO₄ calcined with various calcination temperatures at 500, 600, 700, 800, and 900 °C. Co,N Bidoped ZrO₂ was synthesized by reflux method as control process. All synthesized materials were characterized by X-Ray diffraction (XRD), scanning electron microscopy with energy dispersive X-ray (SEM-EDX), fourier transform infrared spectrophotometer (FTIR), and specular reflectance UV-Visible spectrophotometer (SRUV).

The result obtained showed that Co,N *Bidoped* ZrTiO₄ was successfully synthesized by sol-gel methode. XRD Characterization showed that crytal phase of photocatalysts were dominated by tetragonal and anatase at 500 °C calcination temprerature. ZrO₂ as supporting material enhanced termal stability of Co,N bidoped ZrTiO₄. SEM-EDX analysis showed dopan Co and N successfully doped into ZrTiO₄ framework. IR spectra showed intensity change in the absorption of Ti-O bond. SRUV analysis showed that Co and N dopan reduced band gap of TiO₂. Material with 3 wt.-% of Co concentration calcined at 500 °C has the lowest band gap (2.54 eV) and the longest absorption edge wavelength (487 nm).

Keywords: Co,N *Bidoped* ZrTiO₄, photocatalyst, sol-gel, calcination temperature, various concentration