

## **SINTESIS TURUNAN KHALKON DARI 2,4-DIHIDROKSIASETOFENON DENGAN VANILIN TERHALOGENASI DAN UJI AKTIVITASNYA SEBAGAI ANTIOKSIDAN**

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### **INTISARI**

Sintesis dan uji aktivitas antioksidan telah dilakukan untuk 3-kloro-4-hidroksi-5-metoksibenzaldehida (senyawa **1**), 3-bromo-4-hidroksi-5-metoksibenzaldehida (senyawa **2**), 2',4'-dihidroksi-3-kloro-4-hidroksi-5-metoksihalkon (senyawa **3**) dan 2',4'-dihidroksi-3-bromo-4-hidroksi-5-metoksihalkon (senyawa **4**). Senyawa **1** diperoleh melalui reaksi klorinasi antara vanilin dengan  $\text{Ca}(\text{ClO})_2$ . Senyawa **2** diperoleh melalui reaksi brominasi antara vanilin dengan  $\text{KBrO}_3$  dalam pelarut asam asetat glasial dan katalis  $\text{HBr}$ . Senyawa **3** dan **4** diperoleh dari senyawa **1** dan **2** yang direaksikan dengan 2,4-dihidroksiasetofenon melalui kondensasi Claisen-Schmidt dengan menggunakan katalis basa  $\text{KOH}$  40% dan KSF montmorillonit dalam pelarut metanol. Semua produk dilakukan karakterisasi menggunakan FTIR, GC-MS, TLC-Scanner, MS-Direct,  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR. Senyawa **1**, **2**, **3** dan **4** diuji aktivitas antioksidannya menggunakan DPPH. Hasil penelitian menunjukkan bahwa senyawa **1**, **2**, **3** dan **4** berhasil disintesis dengan rendemen masing-masing sebesar 52,77, 97,00, 1,86, dan 1,00 %. Hasil uji aktivitas antioksidan senyawa **1**, **2**, **3** dan **4** memiliki nilai  $\text{IC}_{50}$  sebesar 244,11, 269,44, 60,00 dan 162,10  $\mu\text{g/mL}$ . Berdasarkan nilai  $\text{IC}_{50}$  senyawa **1** dan **2** tidak memiliki aktivitas antioksidan terhadap DPPH. Senyawa **3** memiliki aktivitas dengan kategori kuat dan senyawa **4** memiliki aktivitas dengan kategori lemah terhadap DPPH.

Kata kunci: klorinasi, brominasi, khalkon, antioksidan, DPPH

***SYNTHESIS OF CHALCONE DERIVATIVES FROM 2,4-  
DIHYDROXYACETOPHENONE WITH HALOGENATED  
VANILLIN AND TESTING ITS ACTIVITY  
AS AN ANTIOXIDANT***

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**ABSTRACT**

The syntheses and antioxidant activity test have been conducted for 3-chloro-4-hydroxy-5-methoxybenzaldehyde (compound **1**), 3-bromo-4-hydroxy-5-methoxybenzaldehyde (compound **2**), 2',4'-dihydroxy-3-chloro-4-hydroxy-5-methoxychalcone (compound **3**) and 2',4'-dihydroxy-3-bromo-4-hydroxy-5-methoxychalcone (compound **4**). Compound **1** was synthesized through a chlorination reaction between vanillin and  $\text{Ca}(\text{ClO})_2$ , Compound **2** was obtained by a bromination reaction between vanillin and  $\text{KBrO}_3$  in glacial acetic acid solvent with  $\text{HBr}$  catalyst. Compounds **3** and **4** were synthesized from compounds **1** and **2** which were reacted with 2,4-dihydroxyacetophenone by Claisen-Schmidt condensation using 40%  $\text{KOH}$  as a base catalyst and KSF montmorillonite in methanol. All synthesized compounds were characterized using FTIR, GC-MS, TLC-Scanner, MS-Direct,  $^1\text{H}$ -NMR and  $^{13}\text{C}$ -NMR. These compounds were also assessed for their antioxidant properties using DPPH. The results showed that compound **1**, **2**, **3** and **4** were successfully synthesized with 52.77%, 97.00%, 1.86%, and 1.00% of yield, respectively. Based on the results of the antioxidant activity test, compounds **1**, **2**, **3** and **4** showed  $\text{IC}_{50}$  values of 244.11, 269.44, 60.00 and 162.10 g/mL, respectively. The  $\text{IC}_{50}$  values of compounds **1** and **2** suggested that these compounds did not have antioxidant activities against DPPH. Meanwhile, compound **3** has a strong category of activity and compound **4** has a weak category of activity against DPPH

**Keywords:** chlorination, bromination, chalcones, antioxidants, DPPH